

# MATERIAL SAFETY DATA SHEET

*This Material Safety Data Sheet meets or exceeds the requirements of the Canadian Controlled Product Regulations (WHMIS)*

## 1. Product and Supplier Identification

**Product:** Patinaetch- Amber

**Product Use:** Concrete stain

**Supplier:** Smart Surface Technology Inc.,  
Unit 143 – 14273 Knox Way,  
Richmond, B.C., Canada, V6V 2Z7  
Telephone: (604) 244-3122  
Facsimile: (604) 244-3115

**Manufacturer:** As above

## 2. Composition

Component	% (w/w)	Exposure Limits
Hydrochloric acid (CAS No.7647-01-0)	1.7-2.1	OSHA PEL-TWA 5ppm ceiling ACGIH TLV-TWA 5ppm ceiling
Ferrous Sulphate (CAS No.7720-78-7)	16-18	OSHA PEL-TWA 1 mg/m <sup>3</sup> for iron dusts and mists as Fe ACGIH TLV-TWA 1 mg/m <sup>3</sup> for iron dusts and mists as Fe
Sodium dichromate (CAS No 10588-01-9)	<1.0	OSHA PEL-TWA 0.1 mg/m <sup>3</sup> ceiling for chromates, as CrO <sub>3</sub> ACGIH TLV-TWA 0.05 mg/m <sup>3</sup> as Cr

## 3. Hazards Identification

**Routes of Entry:** (under normal conditions of use)

Skin Contact: Moderate Eye Contact: Major Ingestion: Moderate Inhalation: Moderate

### **EMERGENCY OVERVIEW:**

Corrosive. The severity of damage depends on the duration of the exposure. In general, solutions and mists with a pH of 3 or less are a significant health concern. Contact with alkali liquids will generate heat. Contact with most metals will generate flammable hydrogen gas.

### **Effects of Short-Term (Acute) Exposure:**

**Inhalation:** Vapour or mist in the 50 to 100 ppm range can cause severe nasal irritation, sore throat, choking, coughing and difficulty breathing. Prolonged exposures can cause burns and ulcers to the nose and throat. Severe exposures for a few minutes at 1000 to 2000 ppm can

**Hazards Identification, continued**

cause a life-threatening accumulation of fluid in the lungs called pulmonary edema. Symptoms of pulmonary edema such as shortness of breath may be delayed for 48 hours after exposure.

**Skin Contact:** Contact with liquid can cause irritation and burns. Vapour or mist may cause redness, irritation and burns if contact is prolonged.

**Eye Contact:** Low concentrations of vapour or mist (10 - 35 ppm) can be immediately irritating and result in redness. Concentrated vapour, mist or splashed liquid can cause severe irritation, burns and permanent blindness.

**Ingestion:** Liquid can cause corrosive burns to mouth, throat, esophagus and stomach. Symptoms may include difficulty in swallowing, intense thirst, nausea, vomiting, diarrhea and in severe cases, collapse and death. Small amounts of acid which enter the lungs during ingestion or vomiting (aspiration) can cause serious lung injury and death.

**Effects of Long-Term (Chronic) Exposure:**

Repeated and prolonged exposure to low concentrations of mist or vapour can cause discolouration and damage to tooth enamel, bleeding of the nose and gums, gastrointestinal symptoms, and chronic bronchitis and gastritis. Repeated exposure to low concentrations of liquid, mist or vapour can cause redness, swelling, sensitization, and pain (dermatitis). Metallic taste and garlic breath are signs of selenium absorption. No evidence of carcinogenicity in human studies. This product does not accumulate in the body.

**Medical Conditions Aggravated By Exposure:**

Pre-existing respiratory and skin disorders.

<h2>4. First Aid Measures</h2>
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**Inhalation:** Take precautions to ensure your own safety before attempting rescue. Wear appropriate personal protective equipment and use the 'buddy' system. Remove victim to fresh air. If breathing has stopped, begin artificial respiration, or if the heart has stopped, begin cardiopulmonary resuscitation (CPR) immediately. Oxygen should be administered by a trained person. Ensure victim is completely at rest - allow no physical exertion. Symptoms may be delayed for up to 48 hours. Immediately transport victim to an emergency medical facility.

**Ingestion:** Never give anything by mouth if victim is rapidly losing consciousness, or is unconscious or is convulsing. Have victim rinse mouth thoroughly with water. **Do not induce vomiting.** Have victim drink 300 mL (10 oz.) of water. If milk is available, administer AFTER the water. If vomiting occurs naturally, have the victim lean forward to reduce risk of aspiration. Repeat administration of water. Immediately transport to emergency medical facility.

**Skin Contact:** Avoid direct contact. Wear impervious protective gloves if necessary. Immediately flush contaminated areas with lukewarm, gently running water for at least 20 minutes. Under running water, remove contaminated clothing, shoes, and leather goods such as watchbands and belts. **Do not interrupt flushing** - have emergency vehicle wait if necessary. Transport victim to emergency medical facility. Decontaminate clothing, shoes and leather goods before reuse or discarding.

**Eye Contact:** Immediately flush contaminated eye(s) with lukewarm, gently running water for at least 30 minutes while holding the eyelid(s) open. Take care not to rinse contaminated water into a non-affected eye. Neutral saline solution may be used for flushing if available. **Do not interrupt flushing** - keep emergency vehicle waiting if necessary. If irritation persists, repeat flushing. Transport victim to emergency medical facility.

**General Comments:** Provide general supportive measures (comfort, warmth, rest). Seek medical attention for all exposures except minor instances of inhalation or skin contact. First-aid

### First Aid Measures, continued

procedures should be reviewed by appropriate personnel familiar with hydrochloric acid and its conditions of use in the workplace.

## 5. Fire Fighting Measures

<b>Flash point:</b>	Not Applicable
<b>Autoignition temperature:</b>	Not applicable. See information under "Fire Fighting Instructions"
<b>Lower Explosive Limit:</b>	Not established
<b>Upper Explosion Limit:</b>	Not established
<b>Sensitivity to Impact:</b>	Not sensitive.
<b>Sensitivity to Static Discharge:</b>	Not sensitive.

**Hazardous Combustion Products:** None. See Hazardous Decomposition Products in Section 10, for information on thermal decomposition.

**Extinguishing Media:** No specific recommendation. Use media to suppress surrounding fire.

**Fire Fighting Instructions:** Wear adequate personal protective equipment. Use water to keep fire-exposed containers cool to prevent rupture. Use water spray or fog to reduce or direct vapours. Do not direct water at source of leak. Trained personnel may neutralize a spill. Contact with common metals produces hydrogen gas that may form explosive mixtures in air.

### NATIONAL FIRE PROTECTION ASSOCIATION (NFPA) HAZARD INDEX:

HEALTH: 3 - Very short exposure could cause serious temporary or residual injury requiring immediate attention.

FLAMMABILITY: 0 - Will not burn.

REACTIVITY: 1 - Normally stable but can become unstable at elevated temperatures and pressures, or may react non-violently with water.

SPECIFIC HAZARDS: Corrosive.

## 6. Accidental Release Measures

**Personal Protection:** Evacuate unnecessary personnel from spill area and keep unprotected persons upwind. Wear appropriate personal protective equipment. Ventilate area. Vapour is heavier than air and will collect in low areas. Do not touch spilled hydrochloric acid.

**Environmental Precautions:** Implement spill control plan. Stop or reduce leak if safe to do so. Prevent from entering sanitary or storm sewers, waterways, or confined spaces. Use inert materials such as earth or sand to form dike.

**Remedial Measures:** Restrict access to area until completion of cleanup. Ensure cleanup is conducted by trained personnel only. Use all appropriate personal protective equipment. For small spills: absorb with neutralizing materials such as soda ash or lime and collect in sealed containers. Flush area with water. For large spills, contain and collect spilled material if possible. Notify government occupational health and safety and environmental authorities as per applicable regulations

## 7. Handling and Storage

**Handling Procedures:** Prevent release of vapour or mist into workplace air. Ensure adequate ventilation. Have emergency equipment readily available. When diluting, slowly add acid to the water to avoid boiling or splattering. Keep containers closed when not in use. Wash face and hands thoroughly after handling and before eating, drinking, or using tobacco products.

**Storage:** Store in a cool, dry, well ventilated area, out of direct sunlight and away from heat sources. Store away from incompatible materials such as oxidizing materials, reducing materials, and strong bases. Keep storage area separate from populated work areas.

## 8. Exposure Controls, Personal Protection

**Engineering Controls:** Use general or local exhaust ventilation to maintain exposure below the exposure limits.

**Respiratory Protection:** If respiratory protection is required, NIOSH recommends for hydrogen chloride in air:

Up to 50 ppm: Chemical cartridge respirator with hydrogen chloride cartridge(s), powered air-purifying respirator with appropriate cartridge(s), Supplied Air Respirator (SAR), or a full face-piece SCBA.

IDLH Conditions (50 ppm) or Planned Entry in Unknown Concentrations: Positive pressure, full face-piece SCBA, or positive pressure full face-piece SAR with an auxiliary positive pressure SCBA.

Escape: Gas mask with canister, or escape type SCBA.

NOTE: Air purifying respirators do not protect against oxygen deficient atmospheres.

**Skin protection:** Wear impervious gloves and boots and/or other protective clothing according to circumstances.

**Eye and Face Protection:** Eye protection is required. Chemical safety goggles are recommended. The wearing of contact lenses is not recommended.

**Footwear:** As required by worksite rules.

**Other:** Have a safety shower and eye wash station readily available in the immediate work area.

## 9. Physical and Chemical Properties

<b>Appearance:</b>	Clear amber liquid ( Amber)	<b>Freezing Point</b>	≈0°C
<b>Odor:</b>	Acrid odour	<b>Boiling Point:</b>	108 °C
<b>Odor Threshold:</b>	Not determined	<b>Critical Temperature:</b>	Not applicable.
<b>pH:</b>	< 1	<b>Relative Density:</b>	≈1.28 (water = 1)
<b>Vapor Pressure:</b>	Not determined	<b>Partition Coefficient:</b>	No data
<b>Solubility:</b>	Completely soluble in water	<b>Evaporation Rate:</b>	Not determined
<b>Vapor Density:</b>	Not determined		